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difference between acids and bases. An acid is a substance that donates an H^+ ion to another chemical species called a base. A base is a substance that accepts (combines with) an H^+ ion.. If you need to know the concentration of the H^+ ion in solution, you can do so by representing the H^+ molarity, $[H^+]$.

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please? A 39.0 mL
sample of 1.05 Moles
KBr and a 56.0 mL
sample of 0.680 Moles

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KBr are mixed. The solution is then heated to evaporate water until the total volume is 65.0 mL.

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A solution is a homogenous mixture that contains two or more substances. Solutions contain a solvent (the substance that dissolves) and a solute (the dissolved

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(substance).

What are ten examples of solutions that you might find in ...

Solution, in chemistry, a homogenous mixture of two or more substances in relative amounts that can be varied continuously up to what is called the limit of solubility. The term solution is commonly applied to the liquid state of

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matter, but solutions of gases and solids are possible.

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solution of 0.02 M acetic acid ($pK_a = 4.76$) and 0.04 M acetate to lower the pH to 2.50? Ignore volume changes. a.

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15. Why is it essential that the denominator for the mass percent calculation must be the total mass of the solution? Because the solute is the minor

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component Because the solvent is the major component of the solution It is not essential So that the percentage of a solute is never greater than 100%

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