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Double

Replacement

Reaction Lab

Answers

# **Double Replacement Reaction Lab Answers**

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### Double

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of chemical reactions.

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## Double

## Replacement

This lab will explore double-replacement reactions, the combination of

atoms/ions reactants that form completely different products. (ex:  $AC + BD \rightarrow AD + BC$ ).

A double replacement takes place between a minimum of two cations and two anions on the reactant side.

These ions produce a minimum of two cations and two anions on the product side.

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## **Double-replacement Reactions**

**ABSTRACT: In this  
lab double ...**

Droplets of reactants such as  $\text{BaCl}_2$  and  $\text{Na}_2\text{SO}_4$  were dropped into spot plates, which created a double replacement reaction. If the substance no longer had an aqueous solution after the double replacement, then the substance would be a precipitate.

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**Displacement**

**Reactions: Forming**

**Precipitate Lab**

**Answers**

10: Double

Replacement Reactions  
(Experiment)

Precipitation Reactions.

Here AB and CD are usually aqueous ionic compounds (or acids) consisting of aqueous ions (A+...

Neutralization

Reactions. Here AB is

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an acid (consisting of  $H^+$  and  $X^-$  aqueous ions) and BC is a base (consisting of  $M^+$ ...

Gas Forming ...

**10: Double Replacement Reactions (Experiment) - Chemistry ...**

Worksheet #5: Double-  
Replacement Reactions

In these reactions, all you do is look at the names of the reactants, and "switch



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partners". Just be sure that the new pairs come out with the positive ion named first, and paired with a negative ion. 1.

aluminum iodide + mercury(II) chloride  $\rightarrow$  aluminum chloride + mercury(II) iodide 2All

## **Worksheet #5: Double-Replacement Reactions In these**

...

Double Replacement  
Pattern:  $AB + CD \rightarrow AD$

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## Replacement

+ CB. Two compounds exchange ions to form two new compounds. In the generic compound "AB", the "A" element, which is usually a metal, exchanges places with another metal in compound "CD". To the right side of the arrow shows their final arrangement.

### **Lab 9: Double Replacement Reactions -**

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**Chemistry Land**

CHM 130LL: Double

Replacement Reactions

One of the main purposes of chemistry is to transform one set of chemicals (the reactants) into another set of chemicals (the products) via a chemical reaction:

Reactants Products

Many of these reactions occur in an aqueous environment (i.e., in a solution where ions and

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compounds

Reaction Lab

## **CHM 130LL: Double Replacement**

### **Reactions**

A double-replacement reaction is a reaction in which the positive and negative ions of two ionic compounds exchange places to form two new compounds. The general form of a double-replacement (also called double-displacement) reaction

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is: (11.9.1)  $AB + CD \rightarrow$   
 $AD + BC$

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**11.9: Double  
Replacement  
Reactions -  
Chemistry  
LibreTexts**

Double replacement reactions lab question? I know from the lab that there are observations to be made to be able to tell if a reaction has occurred such as precipitate forming or

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temperature, etc. But  
the lab question I am  
given is...

Answers

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question? | Yahoo  
Answers**

nhi chung general  
chemistry chem 1411,  
hcc 20 november,  
2017 post lab reactions  
in aqueous solution  
double displacement  
reactions introduction  
the purpose of

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**Post Lab Number  
Eight Reactions in  
Aqueous Solution ...**

Copper(II) sulfate pentahydrate will be dissolved in water and reacted using a double replacement reaction with sodium hydroxide. The addition of hydroxide ions to a solution containing copper(II) ions results in the precipitation of copper(II) hydroxide.

**LAB: Stoichiometry**

*Page 15/24*

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**of a Double**

**Replacement**

**Reaction**

Lab #9 Double

Displacement

Reactions Introduction:

A double displacement

reaction or metathesis

reaction involves the

reaction of two

compounds to form

two new compounds. In

effect, the compounds

change partners with

each other.

**Lab #9 Double**

*Page 16/24*



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**Displacement**

**Reactions Lab**

Answers

The positive ion (cation) from one reactant combines with the negative ion (anion) from the other reactant. The chemical equation shows a double replacement reaction:  $AX + BY \rightarrow AY + BX$  where A and B represent cations, and X and Y represent anions.

**Double Replacement**

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**Lab - CHM130**

**Double Replacement**

**Answers**

DOUBLE REPLACEMENT REACTIONS .

Introduction: You will study double displacement reactions using a small-scale method and predict the products of double displacement reactions. Background: You will combine two water solutions, each containing positive and negative ions. Consider

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this generalized  
reaction between two  
ionic compounds:

Answers

**EXPERIMENT 10:  
DOUBLE  
REPLACEMENT  
REACTIONS  
Introduction**

Pre-Lab Double

Replacement Reactions

Abstract: A double  
replacement reaction is  
when a cation and an  
anion break and form  
new bonds with its  
other counterpart. The

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Answers  
purpose of this experiment is to determine whether a double replacement reaction is occurring based on observation when adding  $\text{Na}_3\text{PO}_4$

**Linmei Amaya -  
Crater High School**

In this Chemthink precipitates lab simulation, you will explore double replacement reactions and precipitate formation. Topics

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Answers  
include: precipitate formation in four different double replacement reactions; writing complete ionic, net ionic, and molecular equations;

Thank you so much to Mr. Charles Sprandal for making this wonderful lab simulation!

**Chemthink\*\*\* -  
Precipitates Lab  
Simulation |  
SimBucket**

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## Double

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Question: FLC Chem  
305 Lab Exercise #7 -

Double Displacement  
Reactions (1) As A

General Rule, All  
Sodium, Potassium,  
And Ammonium

Compounds Are  
Soluble In Water (they  
Not Form Precipitates).

The Same Is True For  
All Nitrate Compounds.

(c) When  $\text{H}_2\text{CO}_3$  Is  
Formed It

Decomposes:  $\text{H}_2\text{CO}_3(\text{g})$

$\text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g})$ . The

Same Is True For

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**Solved: FLC Chem  
305 Lab Exercise #7  
- Double  
Displacement ...**

A double-replacement reaction occurs when parts of two ionic compounds are exchanged, making two new compounds. A characteristic of a double-replacement equation is that there are two compounds as

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reactants and two  
different compounds as  
products.

Answers

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cd98f00b204e9800998  
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